Organic Compounds
Section 1 Simple Organic Compounds Chapter 23 DRW

	<b>Skim</b> Before you read Section 1, skim the headings, illustrations, and captions. Write three questions you have about what may be discussed in this section.
	1
?	2.
	3
Vocabular	Define covalent bond.
covalent bond	
Vocabular	Use your book to define the following key terms.
organic compound	
hydrocarbon	
saturated hydrocarbon	
unsaturated	
hydrocarbon	
isomer	£1 V (1)
isomer .	
	•
benzene	·
·	
Academi	
complex	

### Section 1 Simple Organic Compounds (continued)

### -Main Idea

### Organic Compounds

I found this information on page

### Details

Compare organic and inorganic compounds. Give examples of each type.

Organic Inorganic

Complete the concept map by describing how carbon can form so many organic compounds through bonding and arrangement.

Reasons that carbon can form many organic compounds

### bonding

arrangement

### Hydrocarbons

I found this information on page

**Create** Draw the structural formula for one of the hydrocarbons in this section. Name the hydrocarbon and explain why it is a hydrocarbon on the lines below.

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### Section 1 Simple Organic Compounds (continued)

### Main Idea

#### Bonding in Hydrocarbons

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### Details

Complete the graphic organizer about isomers.

	Isomers have	
identical	different and	different points.

Identify the type of bonds in each hydrocarbon. Make a sketch of each molecule. Then categorize each hydrocarbon as being saturated or unsaturated.

Hydrocarbon	Type of Bond	Sketch	Saturated or Unsaturated
Ethene			
Ethyne			
Propane			
Propene			

CDEATE IT	
CREATE IT Develop a model of a	saturated hydrocarbon using toothpicks a
olored marshmallows. Draw and describe	your model below. Describe how the mod
nows it is a saturated hydrocarbon.	~
/	

# Organic Compounds Section 2 Substituted Hydrocarbons

	Predict Before you read Section 2, look at the headings in the
	section. Write two predictions about what you will learn in this section.
	section.
	1
2.	2
-	
Review Vocabula	PU Define acid.
	boiling acid.
acid	
New	1
Vocabula	Use your book to define each of the following terms.
V COUNTY	To the your book to do not be desired and the years and the years and the years are the years and the years are th
substituted	
hydrocarbon	
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alcohol	
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alcohol	
alcohol	
alcohol ester	
alcohol ester amine	
alcohol ester amine	
alcohol ester amine omatic compound	
alcohol ester amine omatic compound	
alcohol ester amine	Use a dictionary to define framework.
alcohol ester amine omatic compound	C Use a dictionary to define framework.

### Section 2 Substituted Hydrocarbons (continued)

### -Main Idea

#### Replacing Hydrogen

I found this information on page

### Substituting Oxygen Groups

I found this information on page \_\_\_\_\_

#### Substituting Other Elements

I found this information on page \_\_\_\_\_\_

### Details

Evaluate two ways that chemists can change hydrocarbons into other compounds.

- 1.
- Compare alcohols and organic acids using the table below.

How are they formed?	Examples
	How are they formed?

identify five elements other than carbon, hydrogen, and oxygen that can be added to hydrocarbons.

- 2.
- 3.\_\_\_\_\_
- 4.\_\_\_\_\_
- 5. \_\_\_\_\_

### Section 2 Substituted Hydrocarbons (continued)

### Main Idea

### **Aromatic** Compounds

I found this information on page \_

### Details

Complete the table about aromatic compounds.

ies	Definition
es	Structure found in all aromatic compounds
	aromatic compounds

Model the structural formula and symbol for benzene. Place a C at each of the six corners of the symbol, then add the single and double bonds between the carbon atoms. Use the figures in your book to help you.

SUMMARIZE	Write a jo	ournal entry about some organic compounds	you
use. Identify at least of	ne aromatic cor	mpound, one alcohol, and one acid	901
			-

### **Organic Compounds**

Section 3 Petroleum—A Source of Organic Compounds

Scan Use the checklist below to preview Section 3 of your book.

- · Read all section titles.
- · Read all bold words.
- · Read all charts and graphs.
- · Look at all the pictures and read their captions.
- · Think about what you already know about petroleum.

Write two facts you discovered about petroleum as you scanned the section.

1.

2.



Define condense.

condense '



Read the definitions below. Then write the key term on the blank in the left column.

a very large molecule made from small molecules that link together

a small molecule that forms a link in the polymer chain

a process that uses heat or chemicals to break long polymer chains into monomer fragments

Academic Vocabulary

ocabulary Use a dictionary to define link.

link

Name

### Section 3 Petroleum—A Source of Organic Compounds (continued)

### -Main Idea

### What is petroleum?

I found this information on page \_\_\_\_\_\_.

### Processing Crude Oil

I found this information on page \_\_\_\_\_\_.

### Details

**Define** fraction. Define distill. Use a dictionary to help you. Then describe what fractional distillation does to petroleum molecules.

**Sequence** the steps in fractional distillation until the first fraction is separated. One step has been completed for you.

1.

2.

3. The hydrocarbons start to turn into vapor and the vapors rise up inside the tower.

4.

5.

Model a fractionating tower. Use the figure in your book for help. Next to the tower, draw an arrow from the bottom to the top. Along the arrow, show where the higher-boiling fractions condense, the middle-boiling fractions condense, and the lowest-boiling fractions condense.

### Section 3 Petroleum—A Source of Organic Compounds (continued)

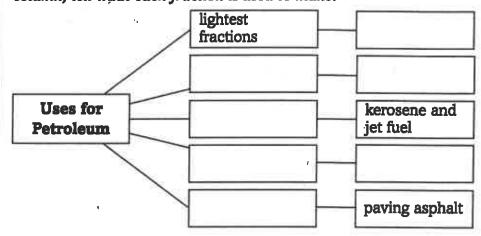
### Main Idea

## Uses for Petroleum Compounds

I found this information on page \_\_\_\_\_\_.

### **Details**

Organize information about the uses of petroleum compounds. In the middle column of boxes, describe the fractions. In the right column, tell what each fraction is used to make.



#### **Polymers**

I found this information on page \_\_\_\_\_.

I found this information on page

Model a polymer if one paper clip is a monomer.

**Describe** at least one benefit and one challenge associated with depolymerization.

Benefit	Challenge

CONNECT IT Consider how your life today would be different without petroleum products. Hypothesize what might be used to replace the petroleum products you use.

Name

Date \_\_\_\_\_

### Organic Compounds

Section 4 Biological Compounds

Skim Before you read Section 4, skim the headings, illustrations, and captions. Write three questions you have about what may be discussed in this section. Review Vocabulary **Define** base. hase locabulary Use your book to define the following key terms. protein carbohydrate lipid nucleic acid deoxyribonucleic acid (DNA) nucleotide Academic ocabulary Use a dictionary to define identical. identical

### Section 4 Biological Compounds (continued)

### -Main Idea

### Details

### Biological Polymers

I found this information on page

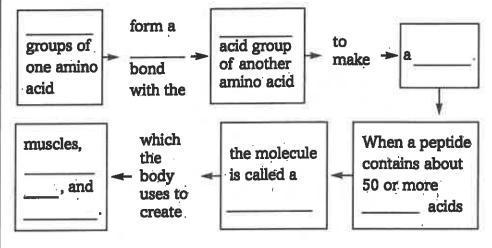
Compare and contrast biological polymers to other polymers.

Similarities	Differences
	ı

### **Proteins**

I found this information on page

Complete the graphic organizer about protein monomers.



### **Carbohydrates**

I found this information on page \_\_\_\_\_\_.

Complete the graphic organizer about carbohydrates.

Carbohydrates,	contain 3 elements:
Starches providefuel.	- provide a quick burst of energy.
Energy from starches can be stored in the cells in the form of	Common table sugar, called, is broken down by digestion into (fruit sugar) and (blood sugar).

Name